M1.	(a)	range of speeds	1	
		moving in different directions accept random motion	1	
	(b)	internal energy	1	
	(c)	density = mass / volume	1	
	(d)	0.00254 / 0.0141	1	
		0.18	1	
		accept 0.18 with no working shown for the 2 calculation marks		
		kg / m ³	1	[7]

M2.	(a)	В		
			no mark for B - marks are for the explanation	
			first two mark points can score even if A is chosen	
		draught in	creases (the rate of) evaporation	
			accept more evaporation happens	
			accept draught removes (evaporated) particles faster	
			do not accept answers in terms of particles gaining energy from the fan / draught	
			nom the lam / draught	1
		evaporation	on has a cooling effect	
			accept (average) <u>kinetic</u> energy of (remaining) particles decreases	
				1
		so temper	ature will fall faster / further	1
	(b)	larger sur	face area	
				1
		increasing	the (rate of) evaporation	
			accept more / faster evaporation	
			accept easier for particles to evaporate	
		or		
		for water t	o evaporate from	
			accept more particles can evaporate	
			accept water / particles which have evaporated are trapped (in the bag)	
			answers in terms of exposure to the Sun are insufficient	

[5]

М3.	(a)	conduction	1
	(b)	 (i) any one from: starting temperature (of cold water) temperature is insufficient 	
		pipe length accept size of pipe	
		pipe diameter	
		pipe (wall) thickness	
		volume of cold water accept amount for volume	
		temperature of hot water (in)	
		• time	1
		(ii) copper	1
		greatest temperature change only scores if copper chosen accept heat for temperature accept heated water the fastest accept it was hottest (after 10 minutes) accept it is the best / a good conductor	1
	(c)	the pipe has a larger (surface) area accept pipe is longer	1
		(so) hot / dirty water (inside pipe) is in contact with cold / clean water (outside pipe) for longer	

1

[6]

M4.(a) Marks awarded for this answer will be determined by the Quality of Written Communication (QWC) as well as the standard of the scientific response. Examiners should also refer to the information in the Marking guidance.

0 marksNo relevant content.

Level 1(1-2 marks)There is a basic explanation of **one** feature**or**a simple statement relating reduction in energy transfer to **one** feature.

Level 2(3-4 marks)There is a clear explanation of **one** feature**or**a simple statement relating reduction in energy transfer to **two** features.

Level 3(5-6 marks)There is a detailed explanation of at least **two** features**or**a simple statement relating reduction in energy transfer to all **four** features.

Examples of the points made in response extra information

accept throughout: heat for energy loss for transfer

plastic cap:

- plastic is a poor conductor
 accept insulator for poor conductor
- stops convection currents forming at the top of the flask so stopping energy transfer by convection
- molecules / particles evaporating from the (hot) liquid cannot move into the (surrounding) air so stops energy transfer by evaporation
- plastic cap reduces / stops energy transfer by conduction / convection / evaporation

glass container:

- glass is a poor conductor so reducing energy transfer by conduction
- glass reduces / stops energy transfer by conduction

vacuum:

• both conduction and convection require a medium / particles

- so stops energy transfer between the two walls by conduction and convection
- vacuum stops energy transfer by conduction / convection

silvered surfaces:

- silvered surfaces reflect infrared radiation accept heat for infrared
- silvered surfaces are poor emitters of infrared radiation
- infrared radiation (partly) reflected back (towards hot liquid)
- silvered surfaces reduce / stop energy transfer by radiation
- (b) (the ears have a) small <u>surface area</u> ears are small is insufficient

so reducing energy radiated / transferred (from the fox)
accept heat lost for energy radiated
do **not** accept stops heat loss

[8]

6

1

M5.	(a)	any two from:	
		water evaporates accept steam / water vapour for water molecules accept water turns to steam	
		water molecules / particles go into the air	
		mirror (surface) is cooler than (damp) air accept the mirror / surface / glass is cold	
		 water molecules / particles that hit the mirror lose energy accept water molecules / particles that hit the mirror cool down 	
		cooler air cannot hold as many water molecules / particles	2
		(causes) condensation (on the mirror) accept steam changes back to water (on the mirror)	
		orparticles move closer together	1
	(b)	mirror (surface) is warm mirror is heated is insufficient	1
		(rate of) condensation reduced accept no condensation (happens)	1

[5]

М6.	(a)	COI	nduction must be in correct order	1
		conv	rection	1
	(b)	<i>(i)</i>	70 accept ± half a square (69.8 to 70.2)	1
		(ii)	15 accept 14.6 to 15.4 for 2 marks allow for 1 mark 70 – 55 ecf from (b)(i) ± half a square	
			col nom (b)(i) 2 mail a cyaalo	2
		(iii)	C	1
			biggest drop in temperature during a given time accept it has the steepest gradient this is a dependent	1
		(iv)	starting at 70 °C and below graph for C must be a curve up to at least 8 minutes	1
		(v)	because 20 °C is room temperature accept same temperature as surroundings	1
	(c)	(i)	6720	

	correct answer with or without working gains 3 marks 6 720 000 gains 2 marks correct substitution of $E = 0.2 \times 4200 \times 8$ gains 2 marks correct substitution of $E = 200 \times 4200 \times 8$ gains 1 mark	3	
(ii)	the fastest particles have enough energy accept molecules for particles	1	
	to escape from the surface of the water	1	
	therefore the mean energy of the remaining particles decreases accept speed for energy	1	
	the lower the mean energy of particles the lower the temperature (of the water) accept speed for energy	1	[16]

M7.Marks awarded for this answer will be determined by the Quality of Written Communication (QWC) as well as the standard of the scientific response. Examiners should also apply a 'best-fit' approach to the marking.

0 marks

No relevant content.

Level 1 (1-2 marks)

Considers either solid or gas and describes at least one aspect of the particles.

or

Considers both solids and gases and describes an aspect of each.

Level 2 (3–4 marks)

Considers both solids and gases and describes aspects of the particles.

or

Considers one state and describes aspects of the particles and explains at least one of the properties.

or

Considers both states and describes an aspect of the particles for both and explains a property for solids or gases.

Level 3 (5–6 marks)

Considers both states of matter and describes the spacing and movement / forces between the particles. Explains a property of both solids and gases.

examples of the points made in the response extra information

Solids

- (particles) close together
- (so) no room for particles to move closer (so hard to compress)
- vibrate about fixed point
- strong forces of attraction (at a distance)
- the forces become repulsive if the particles get closer
- particles strongly held together / not free to move around (shape is fixed)

any explanation of a property must match with the given aspect(s) of the particles.

Gases

- (particles) far apart
- space between particles (so easy to compress)
- move randomly
- negligible / no forces of attraction
- spread out in all directions (to fill the container)

	Correct ariswer only	1
(b)	any two from:	
	 increase the power / watts allow increase the temperature of the oven or make the oven hotter decrease the speed allow leave the biscuits in for longer put biscuits through again increase radiation is insufficient ignore changes to the design of the oven 	2
(c)	(inside) surface is a (good) reflector or poor absorber (of IR) Ignore bounce for reflect surface is a (good) reflector of light does not score surface is a (good) reflector of light and infrared / heat does score	I
	(and) <u>outside</u> surface is poor emitter (of IR)	I
	(so) increases the energy reaching the biscuits allow reduces energy loss or makes oven more efficient do not accept no energy losses keeps oven hotter is insufficient	¹ [6]

M8.

(a)

infrared / IR